

MSE228 Engineering Quantum Mechanics
 Quiz 5 Duration: 30 minutes Open Book Quiz

1. Enumerate all states of the hydrogen atom corresponding to the principal quantum number $n=2$, giving the spectroscopic designation for each. Calculate the energies of these states. How many possible states are there for the $n=3$ level of hydrogen? For the $n=4$ level?

$n=2 \rightsquigarrow l=0,1, \dots, n-1 \rightsquigarrow \underline{l=0,1}$ & $m_l = 0, \pm 1, \pm 2, \dots, \pm l$
 \rightsquigarrow if $l=0 \rightsquigarrow m_l=0$
 if $l=1 \rightsquigarrow m_l=-1, 0, +1$

$n=2 \Rightarrow 2s \text{ \& } 2p$ $2s: n=2 \ l=0 \ m_l=0$ $2p: n=2 \ l=1 \ m_l=-1$ $n=2 \ l=1 \ m_l=0$ $n=2 \ l=1 \ m_l=1$	$\left. \begin{array}{l} \\ \\ \\ \end{array} \right\} \underline{4 \text{ states}}$ $E_n = \frac{E_1}{n^2} (Z^2)$ $= \frac{-13.6 \text{ eV}}{4} = \underline{\underline{-3.4 \text{ eV}}}$
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$n=3 \rightsquigarrow l=0,1,2, m_l=0, \pm 1, \dots, \pm l$
 $n=3 \ l=0 \ m_l=0$
 $l=1 \ m_l=-1 \ m_l=0 \ m_l=1$
 $l=2 \ m_l=-2 \ m_l=-1 \ m_l=0 \ m_l=1 \ m_l=2$

$\left. \begin{array}{l} \\ \\ \end{array} \right\} \underline{9 \text{ states}}$ & $n=3$ & $n=4$
 $\underline{\underline{16 \text{ states}}}$

2. Show that the hydrogen wave function ψ_{211} is normalized.

Hints: $\int_0^\infty r^n e^{-cr} dr = \frac{n!}{c^{n+1}}$ & $\int_0^\pi \sin^3 \theta d\theta = \frac{4}{3}$

$\int |\psi|^2 dv = 1$: normalization
 $\psi_{211} = \frac{1}{8\sqrt{\pi}} \frac{r}{a_0^{3/2}} e^{-r/2a_0} \sin \theta e^{+i\phi}$

$\int \psi_{211}^* \psi_{211} dv = \int \psi_{211}^* \psi_{211} r^2 \sin \theta dr d\theta d\phi$
 $\int \frac{r^2}{64\pi a_0^5} e^{-r/a_0} \sin^2 \theta r^2 \sin \theta dr d\theta d\phi$

$\int_0^\infty \frac{r^2}{64\pi a_0^5} e^{-r/a_0} r^2 dr \int_0^\pi \sin^3 \theta d\theta \int_0^{2\pi} d\phi = 1$

$\frac{1}{64\pi a_0^5} \int_0^\infty r^4 e^{-r/a_0} dr \cdot \frac{4}{3} \cdot 2\pi = 1$

$\frac{4!}{\left(\frac{1}{a_0}\right)^5} \left[\frac{1}{64\pi a_0^5} \cdot \frac{24}{1 a_0^5} \cdot \frac{4}{3} \cdot 2\pi = 1 \right]$
confirmed

$\int_0^\pi \sin^3 \theta d\theta = \int_0^\pi (1 - \cos^2 \theta) \sin \theta d\theta$
 $u = \cos \theta$
 $du = -\sin \theta d\theta$
 $\int (1 - u^2) du \left(\frac{\sin \theta}{-\sin \theta} \right) = -\int (1 - u^2) du$
 $= -\left(u - \frac{u^3}{3}\right) = -u + \frac{u^3}{3}$
 $= -\cos \theta + \frac{\cos^3 \theta}{3} \Big|_0^\pi = \frac{4}{3}$