

MSE228 Engineering Quantum Mechanics
 Quiz 2 Duration: 30 minutes Open Book Quiz

1. Show that de Broglie wavelength of electron with having kinetic energy value of 100 keV is 0.0037 nm.

$KE = 100 \text{ keV} = 0.1 \text{ MeV}$ which is comparable to electron rest mass energy: $0.51 \text{ MeV} \Rightarrow$ Relativistic Equations

$$\textcircled{1} E = KE + mc^2 = KE + E_0$$

$$\textcircled{2} E^2 = (mc^2)^2 + (pc)^2 = E_0^2 + (pc)^2$$

$$pc = h\nu = \frac{hc}{\lambda} : \lambda = ?$$

$$\textcircled{1} \& \textcircled{2} (KE + E_0)^2 = E_0^2 + \left(\frac{hc}{\lambda}\right)^2$$

$$(0.1 \text{ MeV} + 0.51 \text{ MeV})^2 = (0.51 \text{ MeV})^2 + \left(\frac{(6.626 \times 10^{-34} \text{ J}\cdot\text{s}) / (1.602 \times 10^{-19} \text{ J/eV}) (3 \times 10^8 \text{ m/s})}{\lambda}\right)^2$$

$$\Rightarrow \lambda^2 = \frac{[(4.14 \times 10^{-15} \text{ eV}\cdot\text{s}) (3 \times 10^8 \text{ m/s})]^2}{3.6 \times 10^{11} \text{ eV}^2 - 2.5 \times 10^{11} \text{ eV}^2} = \frac{[1.24 \times 10^{-6} \text{ eV}\cdot\text{m}]^2}{1.1 \times 10^{11} \text{ eV}^2} = 1.40 \times 10^{-23} \text{ m}^2$$

$$\Rightarrow \lambda = \sqrt{1.40 \times 10^{-23} \text{ m}^2} = \underline{\underline{0.0037 \text{ nm}}}$$

2. An electron moves in the x direction with a speed of $3.6 \times 10^6 \text{ m/s}$. We can measure its speed to a precision of 1%. With what precision can we simultaneously measure its x coordinate?

x-direction, speed of precision of 1% $\Rightarrow \Delta v_x = v_x \times \frac{1}{100}$

$$\Delta p_x = m \Delta v_x = (9.11 \times 10^{-31} \text{ kg}) (3.6 \times 10^6 \text{ m/s}) = 3.3 \times 10^{-26} \text{ kg}\cdot\text{m/s}$$

$$\rightarrow \Delta x \Delta p_x \geq \hbar \rightarrow \Delta x \geq \frac{\hbar}{\Delta p_x} = \frac{1.05 \times 10^{-34} \text{ J}\cdot\text{s}}{3.3 \times 10^{-26} \text{ kg}\cdot\text{m/s}} = 3.18 \times 10^{-9} \text{ m} = \underline{\underline{3.2 \text{ nm}}}$$

$$(J \equiv \text{kg}\cdot\text{m}^2/\text{s}^2)$$

about 10 atoms in that length